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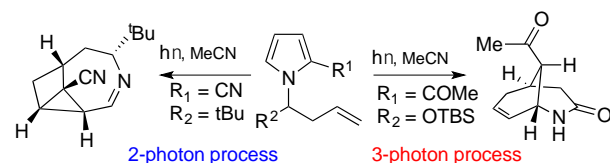
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Conformationally Driven 2- and 3-Photon Cascade Processes in the Stereoselective Photorearrangement of Pyrroles

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Supporting Information Placeholder



ABSTRACT: A TBSO group has been shown to exert a high degree of stereocontrol during the 2-photon photocycloaddition/rearrangement of *N*-butenyl pyrroles to complex tricyclic aziridines. Moreover this and other bulky groups have been shown to change the outcome of the reaction, promoting a new 2-photon sequence to tricyclic imines and an unprecedented stereoselective 3-photon sequence to azabicyclo[3.3.1]nonanes.

Photochemistry is largely unrivalled in its ability to access complex molecular structures from simple starting materials. Part of this is due to reaction types specific to the photochemical excited state such as alkene-alkene [2+2],¹ arene meta-photocycloaddition,² di- π -methine rearrangements³ and various Norrish-Yang⁴ type hydrogen abstraction-recombination sequences to name but a few. Useful sequences involving a second photon mediated step are rare,⁵ but do offer the promise of large increases in molecular complexity due to massive reorganization of the original starting material structure. We recently⁶ described an example of a general 2-photon cascade sequence of pyrroles **1** to give structurally complex aziridines **3** via a second photon mediated rearrangement step of the initially formed cyclobutane **2** (Scheme 1). Herein we describe the unusually high diastereoselectivity observed in photocycloaddition/rearrangement of *N*-butenyl pyrroles bearing bulky groups and how these groups change the course of the reaction to uncover unprecedented 2- and 3-photon mediated processes.

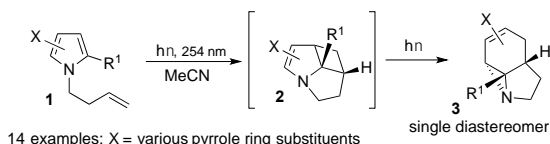
Unlike thermal chemistry, and outside of photoredox catalysis, asymmetric bond formation from photochemically excited states remains highly challenging due to the very short lifetimes involved. Only recently have effective organocatalysts for asymmetric photochemistry been developed as exemplified by the very elegant host-guest sensitizer work of Bach *et al.*⁷ Successful enantioselective photochemical reactions induced by chiral auxiliaries have been reported. However despite notable successes, diastereomeric ratios (*d.r.*) in these auxiliary controlled reactions can often be lower than ideal⁸ as the rapid reaction of excited state intermediates can thwart the establishment of key auxiliary controlled transition states.

Against this backdrop we wished to explore whether functional groups (R^2/R^3) on the butenyl side chain of pyrrole **4** would influence the outcome of this complex photochemical sequence and lead to single diastereomers of sp^3 -rich aziridines as reactive scaffolds⁹ for use in drug discovery. Initially we focused on synthesizing a range of 2-substituted pyrroles

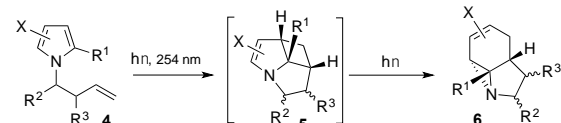
(R^1 = CO-Aux) bearing common chiral auxiliaries such as (+)-menthol, Evans' auxiliary, camphor sultam, borneol, and various chiral amines. All attempts gave aziridines (R^1 = CO-Aux; R^2/R^3 = H) with very little or no observed diastereoselection. This result was surprising given the likely proximity of the butenyl double bond to the 2-acyl unit bearing the chiral auxiliary during the initial [2+2] cycloaddition. Moving to substituents on the butenyl side chain, we initially studied the β -methyl pyrrole **4** (R^1 = CONHEt, R^2 = H, R^3 = Me). Although the aziridine **6** (R^1 = CONHEt, R^2 = H, R^3 = Me) was obtained in 56% yield the *exo/endo* diastereoselectivity was low at 4:1, although still better than the R^1 auxiliary approach (Scheme 1).

Scheme 1. Photocycloaddition-rearrangement reactions of pyrroles

Previous work: two photon rearrangement of pyrroles⁶



This work: influence of R^2/R^3



We therefore decided to study a range of α -substituted pyrrole examples **7**. Moving the methyl group to the α -position (Table 1, entry 1), gave excellent diastereoselectivity ($\geq 98:2$ *dr*) during the two photon process. Pleasingly, this high level of stereocontrol was maintained for a range of α -substituted examples bearing even more bulky alkyl groups (entries 2-5). Interestingly however, on switching from an amide activating group (R^1) to nitrile a significant drop in dia-

stereocontrol was observed which was not improved by switching from Me to Pr (entries 6-7). Use of more functionally useful groups led to interesting results. Although $R^2 = \text{OAc}$ (entry 8) gave very poor dr, the use of $R^2 = \text{OTBS}$ gave high selectivity (8:1). The lower yields obtained with the nitrile activating group are a reflection of the known competing 2- to 3-cyano photo-rearrangement¹⁰ of pyrroles. In all of these examples, NMR and X-ray experiments indicated that the *endo* product (**9**) was the major diastereomer formed, where the R^2 group was placed in the concave face of the 'bowl' generated by the three new rings (*vide infra*).

Table 1: Exploring the effects of stereocontrol with various α -substituents during the photochemical rearrangement of pyrroles

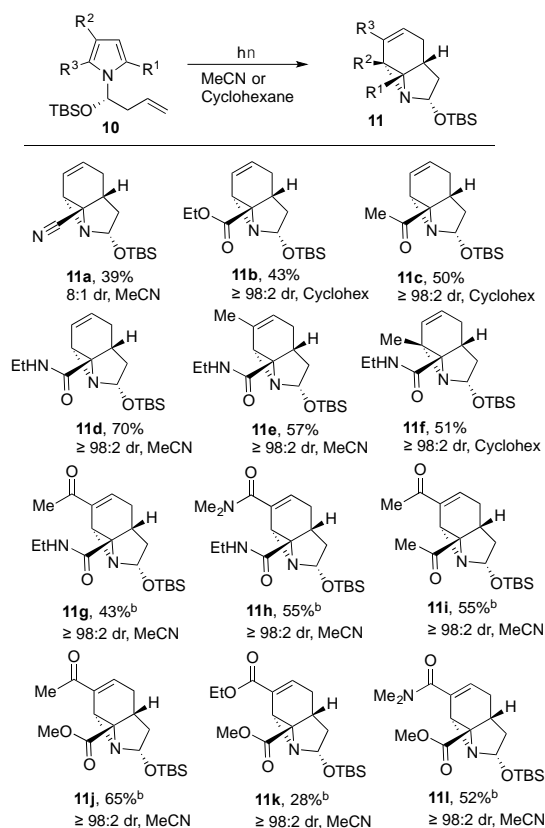
entry	R^1	R^2	9 (%)	dr
1	CONHEt	Me	39	$\geq 98:2$
2	CONHEt	Et	51	$\geq 98:2$
3	CONHEt	Pr	46	$\geq 98:2$
4	CONHEt	iBu	60	$\geq 98:2$
5	CONHEt	iPr	38	$\geq 98:2$
6	CN	Me	37	5:1
7	CN	Pr	40	5:1
8	CN	OAc	50	2:1
9	CN	OTBS	39	8:1
10	CONHEt	OTBS	70	$\geq 98:2$

[a] Reactions performed in MeCN; [b] All products are racemic.

We were keen to exploit this unusually high level of stereoselectivity observed by the OTBS group, especially since the aziridine products would possess a useful functional group (R^2) for further elaboration. The requisite pyrrole carbinols (**7**, $R^2 = \text{OH}$) have been described by Evans as remarkably stable and have been frequently used as masked aldehydes.¹¹ Using this methodology we synthesized a range of substituted pyrroles containing OTBS ethers at the α -substituted carbon and investigated their 2-photon rearrangement (Scheme 2). All 12 examples gave excellent levels of selectivity, with essentially the *endo*-isomer of the aziridines **11a-l** being formed as the single product in each case. In examples possessing two electron withdrawing groups (**11g-l**), better yields were obtained with a dual lamp system at 254 and 312 nm, where the longer wavelength was a better match for the now conjugated cyclobutane intermediates. Only the nitrile example **11a** showed less than optimal selectivity at 8:1 dr. With the exception of **11k** the yields of these rearrangements can be considered moderate to good *i.e.* this is a complex cascade process involving two photons of very high energy (254 nm), generating very strained tricyclic products of high reactivity.^{9,12} As the stereogenic centre bearing the OTBS group had such a profound and consistent level of stereocontrol during rearrangement, we were keen to prove that an asymmetric synthesis of **11** could be achieved in high *e.e.* from enantiopure **10**. Syn-

thesis of *S*-**10d** ($R^1 = \text{CONHEt}$, $R^2 = \text{OTBS}$) gave enantio-enriched **11d** (98% *e.e.*) with no erosion of chirality observed during the photochemical process (see Supporting Information).

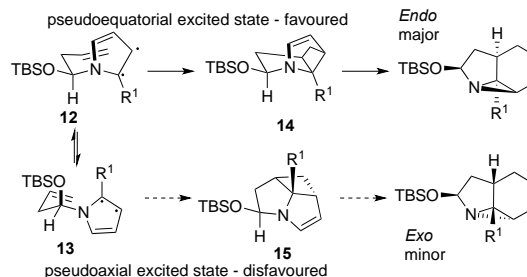
Scheme 2: Scope of the OTBS as a chiral auxiliary in the diastereoselective photochemical rearrangement of pyrroles



[a] All products except **11d** were synthesized as racemates; [b] Concurrent irradiation at 312 nm was employed using a dual lamp system (see SI).

Initially we found it curious that the products were formed as the most hindered isomers *i.e.* the large OTBS group placed on the concave face of the tricyclic aziridine ring system. The origin of this high selectivity is very likely to arise during the first step in the formation of the cyclobutane ring.¹³ If a simple diradical-like mechanism is considered, then after initial excitation a Beckwith radical cyclisation transition state model¹⁴ could be adopted (Scheme 3).

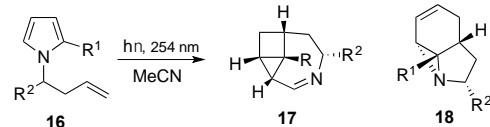
Scheme 3. Stereochemical model to explain *endo*-selectivity during cycloaddition/rearrangement of pyrroles



Here the diradical could adopt one of two equilibrating chair-like transition states **12** and **13**, where the former, pseudoequatorial disposition of the OTBS group would be greatly favoured. On cyclisation of **12** the cyclobutane **14** would result, which in turn would lead to the *endo*-isomer of **11**. This well documented Beckwith model for stereoselective cyclisation neatly explains why in our case the more sterically congested *endo*-isomers of the aziridines are obtained in every case (Scheme 2). This also explains the *exo* selectivity obtained for β -substituted methyl isomer **6** ($R^1 = \text{CONHEt}$, $R^2 = \text{H}$, $R^3 = \text{Me}$).

Whilst attempting to increase the dr of the 2-cyano substituted pyrroles (Table 1, entries 6-9) we explored a range of alkyl substituents of increasing steric bulk. During the course of this we uncovered a novel mode of reactivity that we had not previously observed. For example, irradiation of the *i*Bu substrate **16** (Table 2, entry 1, $R^1 = \text{CN}$, $R^2 = i\text{Bu}$) gave the aziridine **18** ($R^1 = \text{CN}$, $R^2 = i\text{Bu}$) with an improved dr of 11:1. However, along with this we isolated the remarkable tricyclic imine **17** ($R^1 = \text{CN}$, $R^2 = i\text{Bu}$) as a single isomer. We then synthesized a variety of pyrroles with more sterically demanding R^1 groups. In all cases this unusual tricyclic product was observed and indeed with very bulky groups this was the main product (entries 3-5). It is also interesting to note that as the steric bulk of R^2 increased so too did the dr, to the point that no other isomers of the aziridine **18** were detected.

Table 2. Discovery of a new conformationally controlled 2-photon cycloaddition-rearrangement sequence



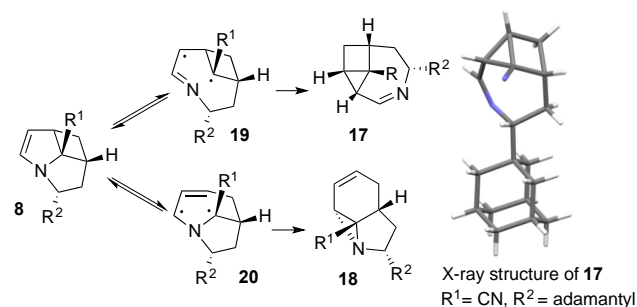
entry	R^1	R^2	17 (%)	18 (%)	dr ^a (18)
1	CN	<i>i</i> Bu	19	35	11:1
2	CN	Cy	8	27	>98:2
3	CN	<i>i</i> Pr	22	7	>98:2
4	CN	<i>t</i> Bu	52	13	>98:2
5	CN	Ad	47	10	>98:2
6 ^b	CO ₂ Et	<i>t</i> Bu	23	29	>98:2

[a] Determined by ¹H NMR; [b] Reaction performed in cyclohexane; [c] All products are racemic.

It is likely that this is a two-photon process proceeding via the initial cyclobutane **8** as a common intermediate. As previously discussed aziridine formation (**18**) is thought to proceed from rearrangement of **8** via the diradical species **20**. Clearly the large R^2 group is exerting a degree of conformational control that favours fragmentation of **8** to the imine diradical **19**, which then cyclises to **17** (Scheme 4). In our previous work we proved that irradiation of intermediate cyclobutanes (**8**) led to aziridine formation and small amounts of the starting pyrrole. However, irradiation of the aziridines showed no sign of the reverse rearrangement to the cyclobutanes. We believe that the formation of **19** likely occurs directly from an excited state of **8**, which influenced by R^2 , adopts a conformation that favours fragmentation to **19**. It is also possible that in **20** the large R^2 now dictates a conformation which disfavours radical recombination to **18** and so equilibrates back to **8**. This 3,4,7-fused ring system is novel and stable and the X-ray structure

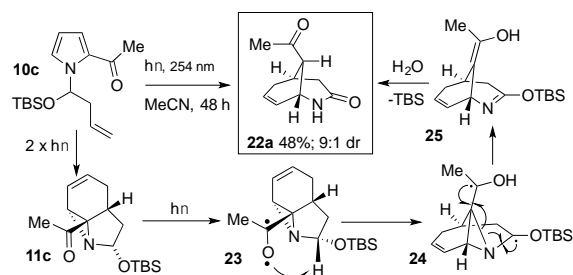
of the crystalline adamantyl derivative **17** ($R^1 = \text{CN}$, $R^2 = \text{Ad}$) shows the interesting fusion of 3,4 and 7-membered rings. The 2-photon process represents an entirely novel photochemical reaction, and further studies are ongoing to determine its generality. In particular, the ability to influence very complicated 2-photon pathways by choice of R^2 group size introduces the exciting prospect of control of short-lived reactive species.

Scheme 4. Mechanism of tricyclic imine formation



During the extensive work with the TBSO derivatives (Scheme 2) we became aware that traces of a new product were being formed in addition to the aziridine. Prolonged irradiation of **10c** showed the appearance of the aziridine **11c**, which over time was consumed to give a new product, the lactam **22a** in 50% yield as essentially a single diastereomer. This remarkable product would appear to be a 3-photon process involving pyrrole→cyclobutane→aziridine→lactam. Further excitation of the ketone in the initially formed aziridine **11c** leads to a Norrish Type II hydrogen atom abstraction sequence **23** to **24**. The resulting biradical **24** then undergoes fragmentation to **25** and upon protonation and desilylation yields **22a** (Scheme 5). This is the first time that we have observed fragmentation of this particular aziridine bond; reactions with nucleophiles, organometallics and internal H-atom transfer process have all proceeded with cleavage of the alternative N-C bond. We believe that the OTBS group performs two crucial roles in directing this alternative pathway. Firstly, the *endo* stereochemistry places the key H-atom in close proximity to the excited ketone in **23**. Secondly the radical centre resulting from abstraction is additionally stabilised by virtue of the α -oxygen atom, explaining why this mode has not been observed in other 2-acyl pyrrole systems.

Scheme 5. Novel three-photon conversion of pyrroles to bicyclic lactams and proposed mechanism

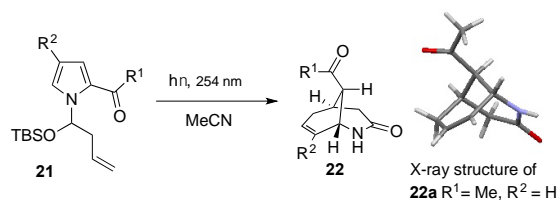


It was interesting to note that enol-keto tautomerisation of **25** would appear to proceed by protonation from the amide/OTBS side of the molecule to give predominantly one diastereoisomer. As the resulting ketone proved stable to epimerisation during isolation and purification by flash chromatography, this is likely a kinetic protonation to the stereoi-

somer observed. Although we were never able to observe the silyloxy-imine motif in **25**, such species are very labile¹⁵ and would likely undergo rapid hydrolysis to the amide in the hygroscopic MeCN solvent used.

An optimization study showed that a ketone in the 2-position of the pyrrole was essential to enable lactam formation (Table 3, entries 1-5) and that a second ketone was actually deleterious to the process (entry 6). Although yields averaged around 50%, this in itself is quite remarkable as intense, high energy 254 nm UV light is used and the successful formation of **22** relies on 3-successive and independent photon mediated events. Unfortunately three sequential photochemical reactions make for a very low overall quantum yield and thus proved to be a scalability issue in batch. However, we were keen to make gram quantities of these compounds and were pleased to be able to produce 1.83 g of **22c** ($R^1 = \text{Cy}$, $R^2 = \text{H}$) in a 13 h run in our 3 x 36W 254 nm FEP flow reactor. This highlights once again the value of flow photochemistry for scaling up very unproductive batch reactions.¹⁶

Table 3. Discovery of a novel 3-photon rearrangement cascade sequence of 2-acylpyrroles to 2-azabicyclo[3.3.1]nonanes



entry	product	R^1	R^2	22 (%)	dr ^[a]
1	22a	Me	H	43	9:1
2	22b	Et	H	53	97:3
3	22c	Cy	H	57 (54) ^c	98:2
4	22d	<i>t</i> Bu	H	39	98:2
5	22e	$\text{CH}_2\text{C}(\text{Me})_2\text{Ph}$	H	41	92:8
6	22f	Me	Ac	31	97:3

[a] Determined by ¹H NMR; [b] Products are racemic; [c] Flow reaction yield

It has been demonstrated that in the photocycloaddition-rearrangement of simple pyrroles, OTBS substitution of the *N*-butenyl side chain exerts powerful stereocontrol during subsequent tricyclic aziridine formation. Further investigation of 2-cyano pyrroles bearing other bulky groups led to the discovery of a novel 2-photon pathway for the formation of highly unusual 3,4,7-fused imine ring systems. In the case of 2-acyl pyrroles, OTBS substitution facilitates an unprecedented and highly stereoselective 3-photon cascade sequence leading to azabicyclo[3.3.1]nonanes. Overall this further underlines the versatility of photochemistry for the synthesis of highly complex molecules from simple starting materials. The ability to select a desired reaction manifold through substituent choice allows access to broad areas of molecular space from a common structural starting point, potentially making it of considerable value in drug discovery.

Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: xxxxxxxx. Synthesis procedures; additional spectral and characterization data, including ¹H, ¹³C NMR (PDF) and X-ray data for **17** and **22a** (CIF)

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